



Abundance of H and He

We can use the **ideal gas** argument of Lecture 3 to estimate the temperature required for hydrogen and helium to escape from a planetary atmosphere:-

$$\begin{aligned}
 T_{\text{esc}} &= \frac{1}{54} \frac{GM_p m}{k R_p} = \frac{1}{54} \frac{GM_p \mu m_H}{k R_p} \\
 &= \frac{6.673 \times 10^{-11} \times 5.976 \times 10^{24} \times 1.674 \times 10^{-27} (M_p / M_{\text{Earth}}) \mu}{54 \times 1.381 \times 10^{-23} \times 6.378 \times 10^6 (R_p / R_{\text{Earth}})} \text{ K} \\
 &= \frac{140 (M_p / M_{\text{Earth}}) \mu}{(R_p / R_{\text{Earth}})} \text{ K}
 \end{aligned}$$